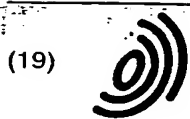


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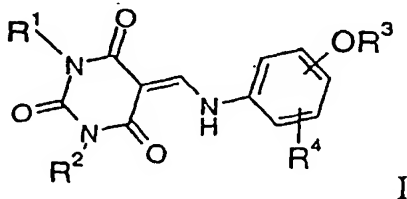
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(54) Novel Pyrimidine-2,4,6-trione compounds

(57) Compounds of the general formula I be used as sunscreen agents.



wherein

R<sup>1</sup> and R<sup>2</sup> each independently are C<sub>2</sub>-C<sub>18</sub> alkyl or C<sub>5</sub>-C<sub>7</sub> cycloalkyl, or C<sub>2</sub>-C<sub>18</sub> alkyl in which at least one methylene group is replaced by oxygen;

R<sup>3</sup> is hydrogen, C<sub>1</sub>-C<sub>18</sub> alkyl or C<sub>2</sub>-C<sub>18</sub> alkyl in which at least one methylene group is replaced by oxygen, or is C<sub>1</sub>-C<sub>18</sub> alkyl carbonyl, C<sub>3</sub>-C<sub>18</sub> alkenyl, C<sub>3</sub>-C<sub>18</sub> alkynyl or a group YS;

R<sup>4</sup> is hydrogen, C<sub>1</sub>-C<sub>8</sub> alkyl or a group OR<sup>3</sup>;

Y is a linker group;

S is a silane-, an oligosiloxane- or a polysiloxane-moiety.

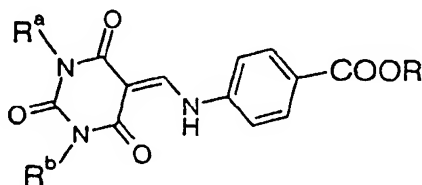
are effective in absorbing ultra violet radiation and can

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## Description

[0001] The present invention relates to novel pyrimidine-2,4,6-trione compounds which are effective in absorbing ultra violet radiation and to light screening compositions comprising said pyrimidine-2,4,6-trione compounds.

[0002] Light screening compositions comprising pyrimidine-2,4,6-trione compounds are described in the International Publication WO 98/14423. This publication refers inter alia to compounds of the general formula



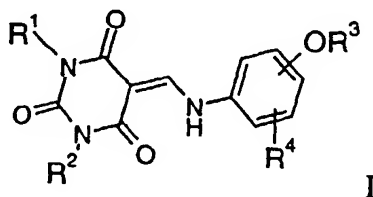
wherein

R is alkyl, cyclic alkyl, aralkyl or aryl;  
R<sup>a</sup> and R<sup>b</sup> each independently are hydrogen, alkyl, cyclic alkyl, aralkyl and aryl.

[0003] The quoted International Publication teaches that these claimed compounds have an excellent ultraviolet absorbing ability. However, these compounds are sensitive to hydrolytic decomposition.

[0004] It has now been found that compounds of the above formula but wherein the group -COOR is replaced by an ether group -OR or an ester group -OCOR exhibit superior chemical stability.

[0005] Thus, the invention relates to compounds of the general formula I



wherein

R<sup>1</sup> and R<sup>2</sup> each independently are C<sub>2</sub>-C<sub>18</sub> alkyl, C<sub>5</sub>-C<sub>7</sub> cycloalkyl or C<sub>2</sub>-C<sub>18</sub> alkyl in which at least one methylene group is replaced by oxygen;

R<sup>3</sup> is hydrogen, C<sub>1</sub>-C<sub>18</sub> alkyl or C<sub>2</sub>-C<sub>18</sub> alkyl in which at least one methylene group is replaced by oxygen, or is C<sub>1</sub>-C<sub>18</sub> alkyl carbonyl, C<sub>3</sub>-C<sub>18</sub> alkenyl, C<sub>3</sub>-C<sub>18</sub> alkynyl or a group YS;

R<sup>4</sup> is hydrogen, C<sub>1</sub>-C<sub>8</sub> alkyl or a group OR<sup>3</sup>;

Y is a linker group;

S is a silane-, an oligosiloxane- or a polysiloxane moiety.

[0006] The invention further relates to light screening compositions containing a compound of formula I, and to the use of a compound of formula I as an agent for absorbing ultraviolet light.

[0007] In the compounds of formula I, R<sup>1</sup> and R<sup>2</sup> can be the same or different. Preferably R<sup>1</sup> and R<sup>2</sup> have the same meaning and are ethyl, propyl, isopropyl, n-butyl, sec. butyl, tert. butyl, 2-ethyl hexyl or cyclohexyl. Particularly preferred is isopropyl.

[0008] The residue OR<sup>3</sup> is preferably in the para position. R<sup>3</sup> preferably is C<sub>2</sub>-C<sub>8</sub> alkyl or a group YS, more prefer-

ably 2-ethyl hexyl or a group YS.

[0009] The residue  $R^4$  preferably is hydrogen or hydroxy.

[0010] The term " $C_1$ - $C_{18}$  alkyl" refers in the present context to straight chain or branched saturated hydrocarbon residues with 1 to 18 carbon atoms such as methyl, ethyl, propyl, isopropyl, thexyl, (1,1,2 dimethylpropyl), n-butyl, sec-butyl, tert. butyl, pentyl, neopentyl, hexyl, 2-ethyl-hexyl, octyl and the like.

[0011] The term " $C_5$ - $C_7$  cycloalkyl" refers in this context to saturated hydrocarbon rings with 5 to 7 carbon atoms i.e. cyclopentyl, cyclohexyl, cycloheptyl.

[0012] The term " $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen" refers in the present context to straight chain or branched saturated hydrocarbon residues with up to 17 carbon atoms which are bound via a carbon atom and have at least one group such as  $-(CH_2-O)-$ ,  $-(CH_2-CH_2-O)-$ ,  $-(CH_2-CH_2-CH_2-O)-$ ,  $-(CH_2-CH_2-CH_2-CH_2-O)-$  and the like.

[0013] In the term " $C_1$ - $C_{18}$  alkyl carbonyl" the alkyl residue is as defined above.

[0014] The term " $C_3$ - $C_{18}$  alkenyl" refers in this context to straight chain or branched unsaturated hydrocarbon residues with 3 to 18 carbon atoms containing at least one double bond such as propen-2-yl, propen-3-yl, buten-3-yl, buten-4-yl, penten-4-yl, penten-5-yl and the like.

[0015] The term " $C_3$ - $C_{18}$  alkynyl" refers in this context to straight chain or branched unsaturated hydrocarbon residues with 3 to 18 carbon atoms containing at least one triple bond. A preferred alkynyl residue is propargyl.

[0016] The term "linker group" refers in this context to a  $C_3$ - $C_{12}$  divalent alkylene or alkenylene chain which links the UV absorbing chromophoric residue to the silane, oligosiloxane or polysiloxane moiety.

[0017] The term " $C_3$ - $C_{12}$  divalent alkylene chain" refers in this context to straight chain or branched saturated hydrocarbon residues such as 3-propylene, 2-propylene, 2-methyl-3-propylene, 3-butylene, 4-butylene, 4-pentylene, 5-pentylene, 6-hexylene and the like.

[0018] The term " $C_3$ - $C_{12}$  divalent alkenylene chain" refers in this context to unsaturated hydrocarbon residues containing at least one double bond such as 2-propen-2-ylene, 2-propen-3-ylene, 3-buten-3-ylene, 3-buten-4-ylene, 4-penten-4-ylene, 4-penten-5-ylene, (3-methyl)-penta-2,4-dien-4 or 5-ylene, 11-dodecen-11-ylene and the like.

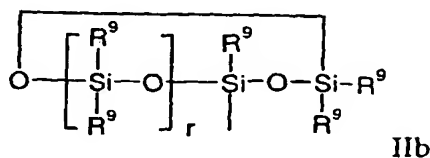
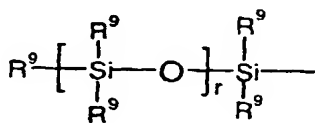
[0019] The divalent alkylene or alkenylene chains may be interrupted by one or several oxygen atoms. Examples for by oxygen interrupted linker chains are e.g. 2-ethoxy-eth-2-ylene, 4-butyloxy-eth-2-ylene or 3,6-dioxa-8-octylene.

[0020] Preferred linker chains are 3-propylene, 4-butylene, 2-propen-2-ylene, 2-propen-3-ylene or 3-buten-4-ylene. Especially preferred is 2-propen-2-ylene or 2-propen-3-ylene.

[0021] The term "silane" refers in this context to a group  $-SiR^5R^6R^7$ , wherein  $R^5$ ,  $R^6$  and  $R^7$  each independently are  $C_1$ - $C_6$  alkyl or phenyl. Preferred silane groups are e.g. trimethylsilane, triethylsilane, tripropylsilane, triisopropylsilane, dimethyl tert.butylsilane, dimethyl thexylsilane, triphenylsilane, dimethylphenylsilane and the like.

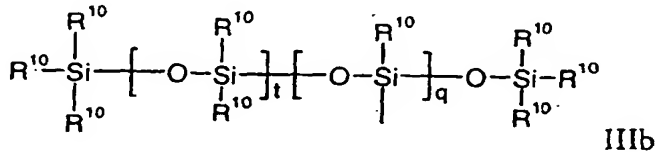
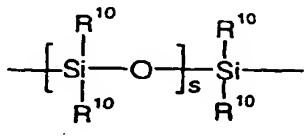
[0022] The term "oligosiloxane" refers in this context to groups of the general formula  $-SiR^8_m(OSiR^8_3)_n$  with  $m = 0, 1$  or  $2$ ;  $n = 1, 2$  or  $3$  and  $m+n = 3$ , wherein  $R^8$  and  $R^8$  each independently are  $C_1$ - $C_6$  alkyl or phenyl. Preferably  $R^8$  and  $R^8$  have the same meaning. Preferred is the group  $-SiMe(OSiMe_3)_2$ , wherein Me signified methyl.

[0023] The term "oligosioxane" further refers to groups of the general formula IIa or IIb



wherein  $R^9$  is  $C_1$ - $C_6$  alkyl or phenyl and  $r$  is an integer from 1 to 9, preferably 1 to 3. The residue  $R^9$  is preferably  $C_1$ - $C_4$  alkyl, more preferably methyl.

[0024] The term "polysiloxane" refers in this context to groups of the general formulae IIIa or IIIb,



wherein

$R^{10}$  is  $C_1$ - $C_6$  alkyl or phenyl;  
 s is an integer from 4 to 250;  
 t is an integer from 5 to 250;  
 q is an integer from 1 to 30.

[0025] Preferably, "s" is an integer from 4 to 150.

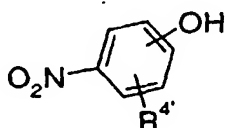
[0026] Preferably, "q" is an integer from 2 to 10, more preferably has statistical mean value of about 4.

[0027] Preferably, "t" is an integer from 5 to 150, more preferably has statistical mean value of about 60.

[0028] The residue  $R^{10}$  is preferably  $C_1$ - $C_4$  alkyl, more preferably methyl.

[0029] The compounds of the general formula I can be prepared by methods known per se.

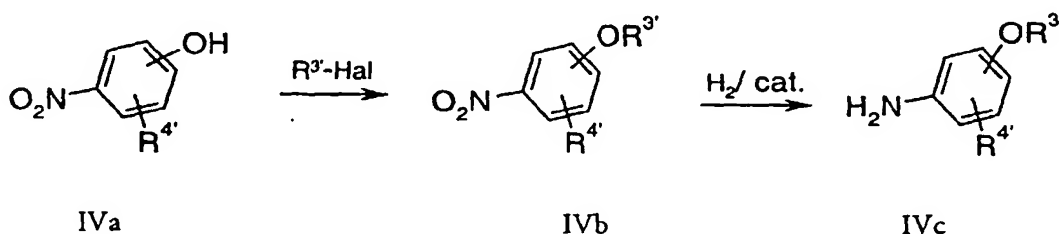
[0030] For compounds of the general formula I, wherein  $R^3$  is  $C_1$ - $C_{18}$  alkyl,  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen,  $C_3$ - $C_{18}$  alkenyl or  $C_3$ - $C_{18}$  alkynyl and  $R^4$  is hydrogen or  $C_1$ - $C_8$  alkyl the process starts by alkylation of a nitrophenol of the formula IVa



wherein

$R^{4'}$  is hydrogen or  $C_1$ - $C_8$  alkyl;

with an alkyl halide to provide a nitro phenolic ether IVb.



#### Step A

[0031] The alkyl halide has the formula  $R^3$ Hal, wherein

$R^3$  is  $C_1$ - $C_{18}$  alkyl,  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen,  $C_3$ - $C_{18}$  alkenyl or  $C_3$ - $C_{18}$  alkynyl; and

Hal is chloro, bromo or iodo, preferably, chloro or bromo.

[0032] The starting nitrophenol is commercially available or may be readily prepared by known methods.

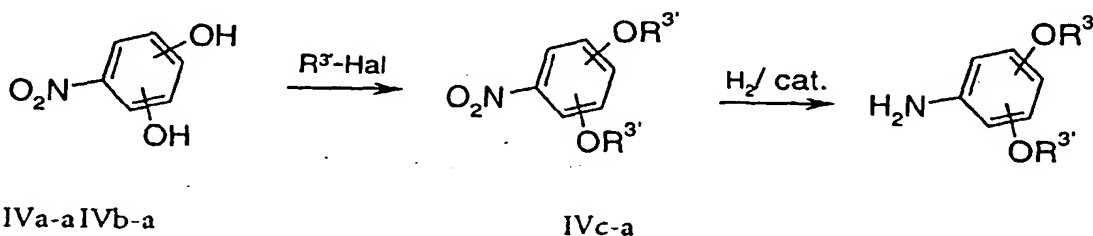
[0033] The reaction is run in a suitable reaction solvent. The choice of the solvent is not critical. Suitable solvents include toluene, pyridine or polar aprotic solvents such as e.g. 1-methyl-2-pyrrolidone, dimethylformamide, dimethylsulfoxide, dioxane, tetrahydrofuran, acetonitrile and the like. The reaction mixture includes an acid acceptor, such as a mild base, for example, sodium or potassium carbonate, which can absorb the acid-by-product of the alkylation. The reactants usually are present in about equal molar amounts and the reaction is run at elevated temperature, e.g. at about  $80^\circ\text{C}$ - $150^\circ\text{C}$ , preferably at about  $100^\circ\text{C}$ .

[0034] The nitro phenolic ether IVb obtained is then reduced either by conventional methods using reduction with tin or zinc and hydrochloric acid or by catalytic hydrogenation to afford the corresponding amino group to provide p-

amino phenolic ether of the formula IVc. The catalytic hydrogenation is employed, using conventional catalysts such as Raney-nickel, palladium or platinum. The catalytic hydrogenation is carried out at temperatures in the range of 0°C to about 100°C, preferably at about 20°C to about 60°C, more preferably at about 40°C and under atmospheric pressure to about 100 bar, preferably at about 20 bar to about 60 bar, more preferably at about 40 bar.

[0035] Nitro phenolic ether compounds wherein  $R^{3'}$  is  $C_3$ - $C_{18}$  alkenyl or  $C_3$ - $C_{18}$  alkynyl are preferably reduced using tin and hydrochloric acid.

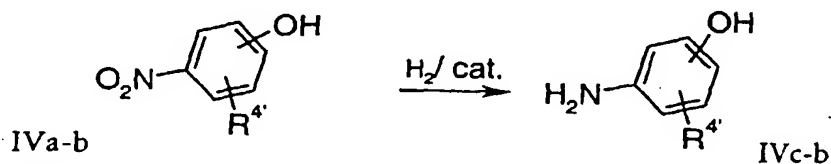
[0036] Compounds of the general formula I, wherein  $R^3$  is  $C_1$ - $C_{18}$  alkyl,  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen,  $C_3$ - $C_{18}$  alkenyl or  $C_3$ - $C_{18}$  alkynyl and  $R^4$  is a group  $OR^{3'}$  can be prepared accordingly.



The group  $R^{3'}$  is as defined above.

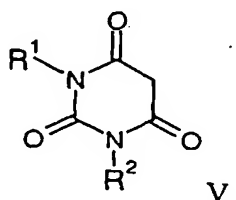
[0037] Compounds of the general formula I, wherein  $R^3$  is  $C_1$ - $C_{18}$  alkyl,  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen,  $C_3$ - $C_{18}$  alkenyl or  $C_3$ - $C_{18}$  alkynyl and  $R^4$  is hydroxy ( $OR^3$  with  $R^3=H$ ) can be prepared using a conventional protective group for the hydroxy group such as e.g. ether-type protective groups or acetal-type protective groups.

[0038] Compounds of the general formula I, wherein  $R^3$  is hydrogen and  $R^4$  is hydrogen or  $C_1$ - $C_8$  alkyl can be prepared by reducing the corresponding nitro phenol IVa-b using conventional methods.

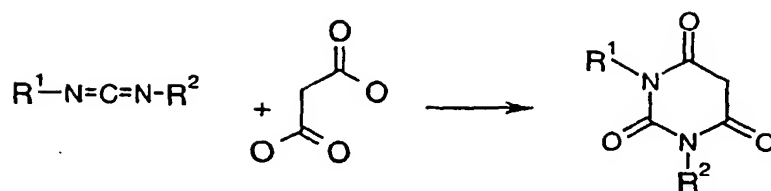


[0039]  $R^{4'}$  is as defined above.

[0040] The pyrimidine-2,4,6-trione residue of the general formula V



wherein  $R^1$  and  $R^2$  are as defined above can be prepared by adding malonic acid in an equimolar quantity to a carbodiimide of the general formula  $R^1-N=C=N-R^2$ , wherein  $R^1$  and  $R^2$  are as defined above. The reaction is run in a suitable reaction solvent. The choice of the solvent is not critical. Preferred are polar aprotic solvents as listed above.

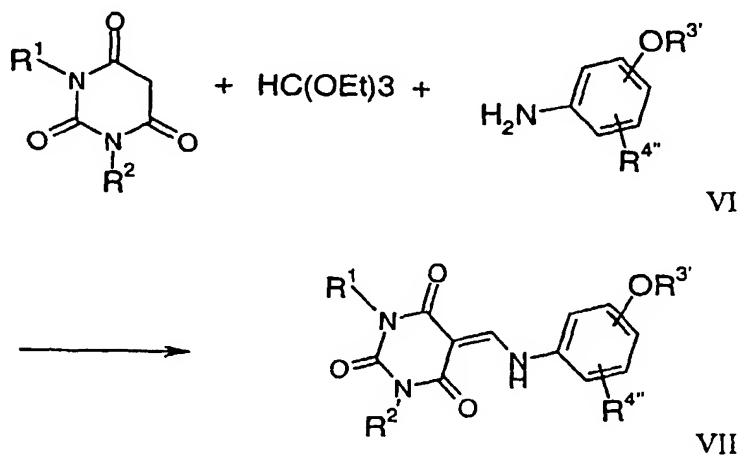


Step B

V

[0041] Other methods to prepare the pyrimidine-2,4,6-trione residue of the general formula V, starting from e.g. malonic ester and dialkyl urea are described in the International Publication WO 95/00112.

[0042] Finally residue V is linked with residue IV by condensation in the presence of trialkyl orthoformate, preferably triethyl orthoformate  $[\text{CH}(\text{OC}_2\text{H}_5)_3]$  to form compound VII

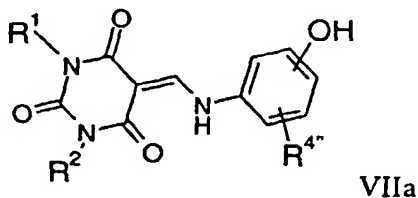


Step C

wherein  $R^1$ ,  $R^2$  and  $R^{3'}$  are as defined above, and  $R^{4''}$  is hydrogen,  $\text{C}_1\text{-C}_8$  alkyl or  $\text{OR}^{3'}$ .

[0043] The condensation is carried out in a solvent at reflux temperatures for several hours. The choice of the solvent is not critical. Preferred are polar aprotic solvents as listed above.

[0044] For preparing a compound of the general formula I it is also possible to first prepared an intermediate of the general formula VIIa

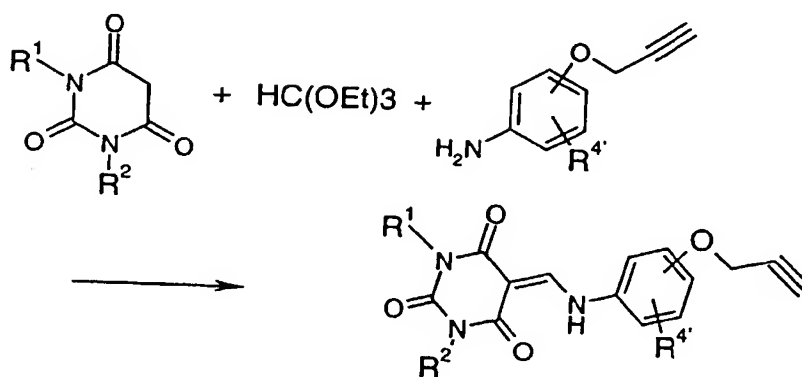
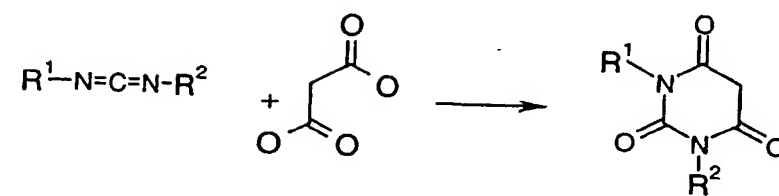
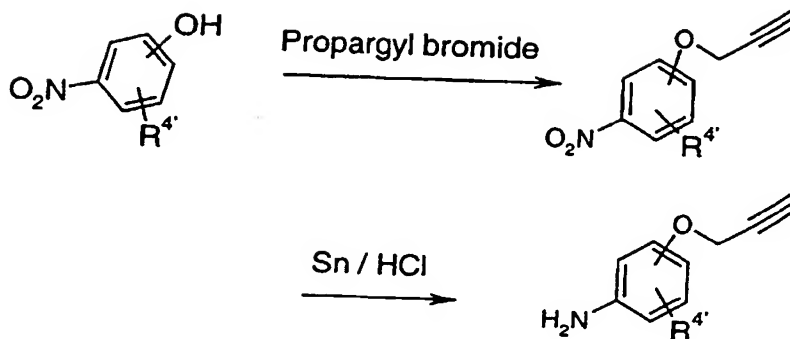


wherein  $R^1$ ,  $R^2$  and  $R^{4''}$  are as defined above by reacting a amino phenol with a pyrimidine 2,4,6-trione compound in

the presence of trialkyl orthoformate, preferably triethyl orthoformate  $[\text{CH}(\text{OC}_2\text{H}_5)_3]$  and then alkylating the resulting intermediate of the formula VIIa.

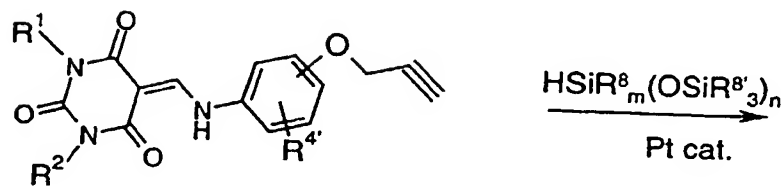
[0045] Compounds of the general formula I, wherein  $\text{R}^3$  is  $\text{C}_1$ - $\text{C}_{18}$  alkyl carbonyl and  $\text{R}^4$  is hydrogen or  $\text{C}_1$ - $\text{C}_8$  alkyl can be prepared by acylating the above compound of the general formula VIIa using known esterification reactions.

5 [0046] Compounds of the general formula I containing a silane-, an oligosiloxane- or a polysiloxane moiety are prepared by hydrosilation. The following reaction scheme shows an example to prepare a compound of the general formula I wherein  $\text{R}^3$  is a group YS wherein S is an oligosiloxane of the formula  $-\text{SiR}^8_m(\text{OSiR}^{8'}_3)_n$  with  $m = 0, 1$  or  $2$ ;  $n = 1, 2$  or  $3$  and  $m+n = 3$ , wherein  $\text{R}^8$  and  $\text{R}^{8'}$  each independently are  $\text{C}_1$ - $\text{C}_6$  alkyl or phenyl and  $\text{R}^4$  is as defined above, namely hydrogen or  $\text{C}_1$ - $\text{C}_8$  alkyl.

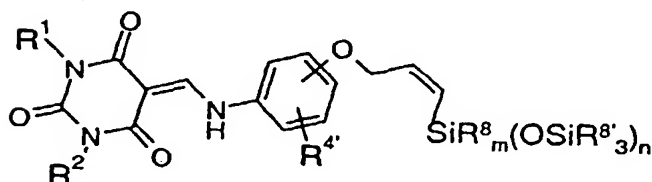
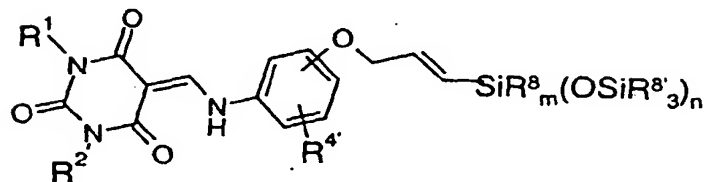
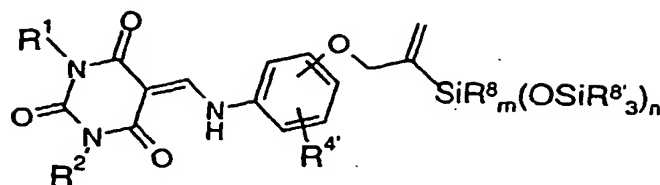


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Step C







### Step D

[0047] The hydrosilation reaction between the SiH group and the alkynyl group (step D) is carried out in the presence of a transition metal catalyst e.g. platinum on charcoal or a platinum complex catalyst such as e.g. divinyl-tetramethyl disioxane platinum complex. The reaction is run in a suitable reaction solvent such as e.g. in toluene. The reactants usually are present in about equal molar amounts and the reaction is run at a somewhat elevated temperature, e.g. at about 60°C-150°C, preferably at about 40°C-100°C, more preferably at about 80°C.

[0048] Compounds of the general formula I containing an oligosiloxane moiety of the general formula IIa or IIb or a polysiloxane moiety of the general formula IIIa or IIIb are prepared accordingly. Reaction partner is the corresponding SiH containing oligosiloxane or polysiloxane moiety.

[0049] The compounds of the general formula I have adsorption maxima in the range of 320 nm to 400 nm, the so called UV-A radiation.

[0050] The preparation of novel light screening agents, especially of preparations for skin protection and, respectively, sunscreen preparations for everyday cosmetics, comprises incorporating a compound of formula I in a cosmetic base which is conventional for light screening agents. Where convenient, other conventional UV-A, and respectively, UV-B filters can also be combined during this incorporation. Said combinations of UV filters can show a synergistic effect. The preparation of said light screening agents is well known to the skilled artisan in this field. The amount of compounds of the general formula I and other known UV-filters is not critical. Suitable amounts are about 0.5 to about 12%.

[0051] Suitable UV B filters, i.e. substances having absorption maxima between about 290 and 320 nm, are for example the following organic compounds which belong to the widest classes of substance:

--- Acrylates such as 2-ethylhexyl 2-cyano-3,3-diphenylacrylate (octocrylene, PARSOL 340), ethyl 2-cyano-3,3-diphenylacrylate and the like;

--- Camphor derivatives such as 4-methyl benzylidene camphor (PARSOL 5000), 3-benzylidene camphor, camphor benzalkonium methosulfate, polyacrylamidomethyl benzylidene camphor, sulfo benzylidene camphor, sulphomethyl benzylidene camphor, therephthalidene dicamphor sulfonic acid and the like;

--- Cinnamate derivatives such as octyl methoxycinnamate (PARSOL MCX), ethoxyethyl methoxycinnamate, diethanolamine methoxycinnamate (PARSOL Hydro), isoamyl methoxycinnamate and the like as well as cinnamic acid derivatives bond to siloxanes:

--Organosioxane compounds containing benzmalonate groups as described in the European Patent Publications EP 0358584 B1, EP 0538431 B1 and EP 0709080 A1;

--- Pigments such as microparticulated  $\text{TiO}_2$ , and the like. The term "microparticulated" refers to a particle size from about 5 nm to about 200 nm, particularly from about 15 nm to about 100 nm. The  $\text{TiO}_2$  particles may also be coated by metal oxides such as e.g. aluminum or zirconium oxides or by organic coatings such as e.g. polyols, methicone, aluminum stearate, alkyl silane. Such coatings are well known in the art.

--- Imidazole derivatives such as e.g. 2-phenyl benzimidazole sulfonic acid and its salts (PARSOL HS). Salts of 2-phenyl benzimidazole sulfonic acid are e.g. alkali salts such as sodium- or potassium salts, ammonium salts, morpholine salts, salts of primary, sec. and tert. amines like monoethanolamine salts, diethanolamine salts and the like.

--- Salicylate derivatives such as isopropylbenzyl salicylate, benzyl salicylate, butyl salicylate, octyl salicylate (NEO HELIOPAN OS), isooctyl salicylate or homomenthyl salicylate (homosalate, HELIOPAN) and the like;

--- Triazone derivatives such as octyl triazone (UVINUL T-150), dioctyl butamido triazone (UVASORB HEB) and the like.

**[0052]** Suitable UV A filters i.e. substances having absorption maxima between about 320 and 400 nm, refers to the following UV-A screening agents.

----Dibenzoylmethane derivatives such as 4-tert. butyl-4'-methoxydibenzoylmethane (PARSOL 1789), dimethoxy-dibenzoylmethane, isopropylidibenzoylmethane and the like;

----Benzotriazole derivatives such as 2,2'-methylene-bis-(6-(2H-benzotriazole-2-yl)-4-(1,1,3,3-tetramethylbutyl)-phenol (TINOSORB M) and the like;

--- Pigments such as microparticulated ZnO and the like. The term "microparticulated" refers to a particle size from about 5 nm to about 200 nm, particularly from about 15 nm to about 100 nm. The ZnO particles may also be coated by metal oxides such as e.g. aluminum or zirconium oxides or by organic coatings such as e.g. polyols, methicone, aluminum stearate, alkyl silane. Such coatings are well known in the art.

**[0053]** As dibenzoylmethane derivatives are photolabile it is necessary to photostabilize these UV-A screening agents. Thus, the term "conventional UV-A screening agent" also refers to dibenzoylmethane derivatives such as e.g. PARSOL 1789 stabilized by the following stabilizing agents:

--- 3,3-Diphenylacrylate derivatives as described in the European Patent Publications EP 0 514 491 B1 and EP 0 780 119 A1;

--- Benzylidene camphor derivatives as described in the US Patent No. 5,605,680;

--- Organosiloxanes containing benzmalonate groups as described in the European Patent Publications EP 0358584 B1, EP 0538431 B1 and EP 0709080 A1.

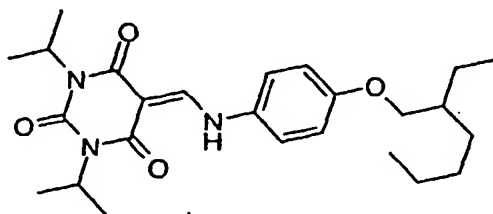
**[0054]** As cosmetic bases conventional for light screening compositions in the scope of the present invention there can be used any conventional preparation which corresponds to the cosmetic requirements, e.g. creams, lotions, emulsions, salves, gels, solutions, sprays, sticks and milks; see also, Sunscreens, Development, Evaluation and Regulatory Aspects, ed. N.Y. Lowe, N.A. Shaath, Marcel Dekker, Inc. New York and Basel, 1990.

**[0055]** Having regard to their good lipophilicity, the compounds of the general formula I can be incorporated well into oil and fat containing cosmetic preparations.

**[0056]** The following examples illustrate the invention in more detail, but do not limit its scope in any manner.

Example 15-[[4-(2-ethyl-hexyloxy)-phenylamine]-methylene]-1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0057]



## a) 4-Nitrophenyl-(2-ethyl-hexyl)-ether

[0058] In a 300 ml three necked reaction flask equipped with a reflux condensor, a thermometer and a magnetic stirrer 27.7g of p-nitrophenol and 52.7g of anhydrous sodium carbonate were suspended in 180 ml 1-methyl-2-pyrrolidone under nitrogen atmosphere. 48 ml of 2-ethyl-hexylbromide were added by means of a dropping funnel. The reaction mixture was heated to 100° C for 180 min. Then the mixture was poured into water and extracted four times with ethyl acetate. The combined organic phases were washed with 1n NaOH and with NaCl solution, dried with Na<sub>2</sub>SO<sub>4</sub> followed by concentration at a rotary evaporator. The product was freed of the residual 2-ethyl-hexylbromide by heating to 60°C at a high vacuum of 0.5 mbar. Yield 32.7g (69% of the theory) of a clear yellow liquid. MS: 192(M<sup>+</sup>), 163, 113, 83, 71, 57(100%).

## b) 4-(2-Ethyl-hexyloxy)-phenylamine

[0059] A 600 ml hydrogenation autoclave was charged with 36 g (0.143 mol) of the above p-nitro phenolic ether and with 5 g of wet Raney Ni catalyst in 50 ml methanol. The reduction was carried out for 18 hours at 40°C and a hydrogen pressure of 40 bar. The mixture was filtered and concentrated, followed by a flash chromatography in hexane/ethyl acetate = 9:1 and "Kugelrohr" distillation (200°C/0.8 mbar). Yield 8.8g of a brownish liquid.

## c) 1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0060] 13.4g of malonic acid was suspended in 130 ml of tetrahydrofuran in a 500ml reaction flask and cooled to 0°C. A solution of 32.6g of N,N'-diisopropyl-carbodiimide in 130 ml of tetrahydrofuran was slowly added within 45 min. at 0°C to 5°C followed by stirring for 2 hours at this temperature. The reaction mixture was concentrated at a rotary evaporator and recrystallized in cyclohexane and then in hexane. Yield 20.6g (75% of the theory) of slightly yellowish crystals, m.p. 145 - 148°C.

## d) 5-[[4-(2-ethyl-hexyloxy)-phenylamine]-methylene]-1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0061] A 100 ml three necked reaction flask equipped with a thermometer a reflux condensor and a magnetic stirrer under nitrogen atmosphere was charged with 1.1g (5 mmol) of 4-(2-ethyl-hexyloxy)-phenylamine (Example 1b), 1.06g (5 mmol) of 1,3-diisopropyl-pyrimidine-2,4,6-trione (Example 1c) and 11.1 g (75 mmol) of triethyl orthoformate in 10 ml of tetrahydrofuran. This mixture was refluxed for 18 hours followed by washing twice with water, dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated at a rotary evaporator. The raw product was recrystallized in ethanol to yield 0.84 g of yellowish crystals. M.p. 74 - 76°C, UV 355 nm (32'122). This product was irradiated in high dilution with a Hg-lamp 150 W from Hereus and has been shown to be photostable.

## e) Measurement of solubility in Cétiol LC (cocoyl caprylate caprate) and Crodamol DA (diisopropyl adipate).

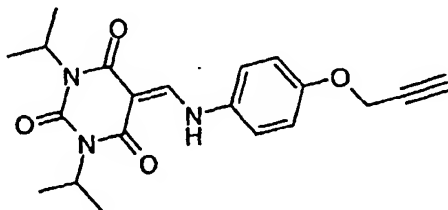
[0062] Oversaturated solutions of 5-[[4-(2-ethyl-hexyloxy)-phenylamine]-methylene]-1,3-diisopropyl-pyrimidine-

2,4,6-trione (see above) in the above solvents were prepared and treated in the ultrasonic bath for five minutes. After standing over night at 25°C the solution was filtered through a microfilter (Millipore, pore size 0.5 µm), followed by UV measurement in CH<sub>2</sub>Cl<sub>2</sub>. The extinctions were compared with the extinctions of the pure compound. The solubilities were found to be 15.5% in Cétiol LC and 26.9% in Crodamol DA.

## Example 2

### 5-((4-propargyloxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0063]



#### a) 4-Nitrophenyl-propargylic-ether

[0064] In a 200 ml three necked reaction flask equipped with a reflux condenser, a thermometer and a magnetic stirrer 20g of p-nitrophenol and 38g of anhydrous sodium carbonate were suspended in 90 ml 1-methyl-2-pyrrolidone under nitrogen atmosphere. 17.3 ml of propargyl bromide were added by means of a dropping funnel. The reaction mixture was heated to 100°C for 60 min. Then the mixture was poured into water and extracted four times with ethyl acetate. The combined organic phases were washed with in NaOH and with NaCl solution, dried with Na<sub>2</sub>SO<sub>4</sub> followed by concentration at a rotary evaporator to form 23.6g of raw product. The product was recrystallized in isopropanol. Yield 21.8g (85% of the theory) of yellowish crystals, m.p. 113-116°C.

#### b) 4-propargyloxy-phenylamin

[0065] A 1 liter four necked reaction flask equipped with a thermometer a reflux condenser and a magnetic stirrer was charged with 21g (118 mmol) of 4-nitrophenyl-propargylicether (Example 2a) in 295 ml of methanol. 295 ml of concentrated hydrochloric acid was slowly added with rigorous stirring, followed by portions of totally 59g of tin powder. After this addition the reaction temperature was kept at 60°C for 2 hours. The mixture was poured into a solution of 295g of sodium hydroxide in 1.2 l of water. Methanol was distilled off at a rotary evaporator. The suspension was filtered and washed with ethanol. The filtrate was concentrated and distilled in the Kugelrohr apparatus at 180°C/0.07 mbar to yield 10.2g of product. This was recrystallized in a little ethanol to yield 6.8g of white crystals of pure product. MS: 147(M<sup>+</sup>), 108 (100%), 80, 53.

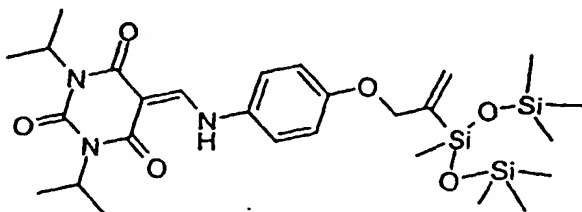
#### c) 5-((4-propargyloxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0066] A 50 ml three necked reaction flask equipped with a thermometer a reflux condenser and a magnetic stirrer under nitrogen atmosphere was charged with 1.45g (10 mmol) of 4-propargyloxy-phenylamin (Example 2b), 2.1g (10 mmol) of 1,3-diisopropyl-pyrimidine-2,4,6-trione (Example 1c) and 2.5 ml of triethyl orthoformate in 15 ml of tetrahydrofuran. This mixture was refluxed for 3.5 hours followed concentrating at a rotary evaporator. The raw product was recrystallized in toluene and hexane to yield 2.9 g of yellow crystals. M.p. 110 - 114°C, UV 352 nm (31'596). This product was irradiated in high dilution with a Hg-lamp 150 W from Hereus and has been shown to be photostable.

Example 3

5-[[4-(2-(1,3,3,3-tetramethyl-1-[(trimethylsilyl)-oxyl-disiloxanyl]-prop-2-enyloxy)-phenylamine)-methylene]-1,3-diisopropyl-pyrimidine-2,4,6-trione]

[0067]

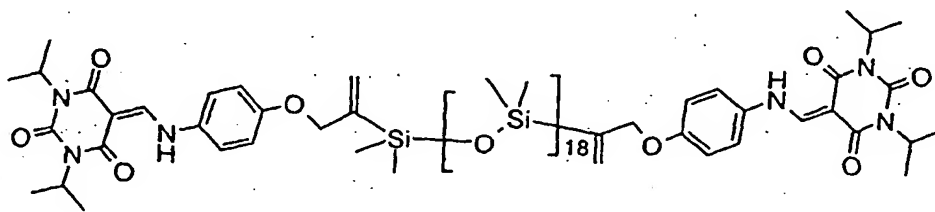


## Hydrosilylation reaction

[0068] 550 mg of the above 5-[[4-(propargyloxy-phenylamine)-methylene]-1,3-diisopropyl-pyrimidine-2,4,6-trione (Example 2c), 330 mg of 1,1,1,3,5,5,5-heptamethyl trisiloxane and a catalytic amount of divinyl-tetramethyl disiloxane platinum complex in 10 ml of toluene was placed in a three-necked reaction flask under inert atmosphere and stirred for 20 hours at 80°C. The product solution was washed with a mixture of water/methanol = 1:10 and concentrated to yield 920 mg. This product was chromatographed in hexane/ethylacetate = 7:1 to yield 650 mg (76% of the theory) of a honey like liquid. UV 354 nm ( $\epsilon=31'520$ ), MS: 591 ( $M^+$ ), 576, 551, 331 (100%) and 221. Its NMR shows a mixture of the vicinal and the geminal hydrosilylation product = 1.3. It has an unlimited solubility in the cosmetic solvents tested above and excellent photostability qualities in high dilution, determined as described in example 1.

Example 4

[0069] A polysiloxane which corresponds in its statistical mean to the following formula:



[0070] 370 mg of the above 5-[[4-(propargyloxy-phenylamine)-methylene]-1,3-diisopropyl-pyrimidine-2,4,6-trione (Example 2c), 810 mg of polysiloxane SLM/4 of Wacker-Chemie GmbH and a catalytic amount of divinyl-tetramethyl disiloxane platinum complex in 10 ml of toluene was placed in a three-necked reaction flask under inert atmosphere and heated for 11 days to 80°C. The product solution was washed with a mixture of water/methanol = 1:10 and concentrated to yield 1170 mg (99%) of a dark liquid. UV 355 nm ( $E=237$ ), having unlimited solubility in Cétol LC and Crodamol DA and excellent photostability qualities in high dilution, determined as described in Example 1.

Example 5

[0071] A polysiloxane which corresponds in its statistical mean to the following formula:



20

5-((4-hydroxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione.

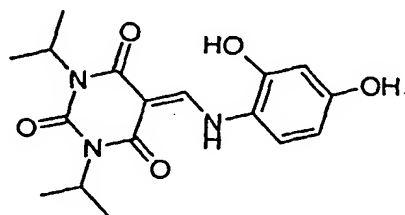
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Example 75-((2,4-dihydroxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione.

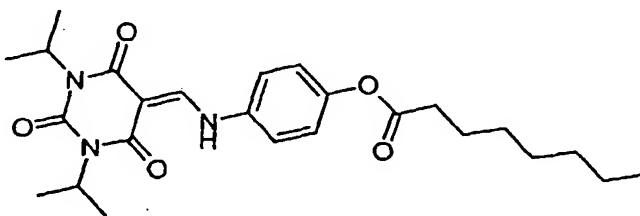
[0075]



A 50 ml three necked reaction flask equipped with a thermometer a reflux condensor and a magnetic stirrer was charged under nitrogen atmosphere with 3.0 g (19 mmol) of 2,4-dihydroxyaniline, 3.9 g (19 mmol) of 1,3-diisopropyl-pyrimidine-2,4,6-trione (Example 1c) and 5.7 g (39 mmol) of triethyl orthoformate in 15 ml of tert. butyl-methyl ether. This mixture was refluxed for 5.5 hours followed by washing twice with water, dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated at a rotary evaporator. The yellow raw product was recrystallized from toluene/hexane to yield 2.4 g of brown crystals. M.p. 225 - 226°C, UV 363 nm (23'564). This product was irradiated in high dilution with a Hg-lamp 150 W from Hereus and has been shown to be photostable. The solubility in Crodamol DA was found to be 4.1 %.

Example 85-((4-capryloxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0076]

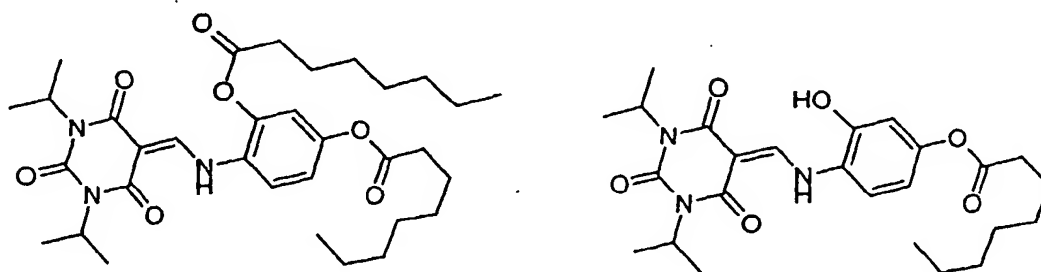


[0077] A 50 ml three necked reaction flask equipped with a thermometer a reflux condensor and a magnetic stirrer was charged under nitrogen atmosphere with 1.66 g (5 mmol) of 5-((4-hydroxy-phenylamine)-methylene)-1,3-diisopropyl-pyrimidine-2,4,6-trione (see Example 6) and 0.48 g (5 mmol) of pyridine in 25 ml of dichloromethane. Then 0.9 g (5.5 mmol) of caprylic acid chloride was slowly added. The reaction started exothermically. The reaction mixture was later heated to 75°C for 22 hours. The mixture was washed twice with each of 1N HCl, NaCl solution and water and back-washed with dichloromethane, dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated at a rotary evaporator. The yellow slowly crystallizing oil was recrystallized from hexane to yield 3.2 g of yellowish crystals (70% of the theory). M.p. 74 - 76°C, UV 346 nm (33'768). This product was also irradiated in high dilution with a Hg-lamp 150 W from Hereus and has been shown to be photostable. The solubility in Crodamol DA was found to be 18.5 % and in Cétol LC 6.6%.

Example 9

5- $\{[(2,4\text{-Bis-capryloxy})\text{-phenylamine})]\text{-methylene}\}$ -1,3-diisopropyl-pyrimidine-2,4,6-trione and 5- $\{[(2\text{-hydroxy-4-capryloxy})\text{-phenylamine})]\text{-methylene}\}$ -1,3-diisopropyl-pyrimidine-2,4,6-trione.

[0078]



[0079] The reaction of Example 8 was repeated using the product of Example 7 (5- $\{[(2,4\text{-dihydroxy-phenylamine})]\text{-methylene}\}$ -1,3-diisopropyl-pyrimidine-2,4,6-trione). The resulting mixture was chromatographed in hexane/methyl-tert.butyl ether to yield two products, the diester in 85% yield, which was a yellowish honey, UV 351 nm (31'349) and easily mixable with Cétiol LC in any proportion and the crystalline 5- $\{[(2\text{-hydroxy-4-capryloxy})\text{-phenylamine})]\text{-methylene}\}$ -1,3-diisopropyl-pyrimidine-2,4,6-trione in 15% yield, m.p. 115-117°C, UV 354 nm (25'813).

Example 10

[0080] Broad spectrum sunscreen lotion containing 2% of a compound of Example 1.



wt%	Compound	INCI name	Supplier
<b>Part A</b>			
2.00	PARSOL MCX	Octyl methoxycinnamate	1
2.00	Product of Example 1		
3.00	PARSOL 1789	4-tert. Butyl-4'-methoxy-dibenzoyl-methane	1
12.00	CETIOL LC	Coco-caprylate/caprate	8
4.00	DERMOL185	Isostearyl neopentanoate	9
0.25	Diethyleneglycolmonostearate	PEG-2-stearate	4
1.00	Cetylalcohol	Cetylalcohol	
0.25	MPOB/PPOB	Methyl-propylparabene	
0.10	EDTA BD	EDTA-sodium salt	5
1.00	AMPHISOL DEA	Diethanolamine cetylphosphate	1
<b>Part B</b>			
20.00	PERMULENE TR-1 (+%)	Acrylate C10-C30 Alkylacrylate	7
48.60	Deionized Water	Deionized Water	
5.00	Propyleneglycol	1,2-Propanediol	
0.80	KOH (10%)	Potassium hydroxide	
INCI: International Nomenclature Cosmetic ingredients			

[0081] Part A was heated in a reactor to 85°C. Part B was slowly added within 10 min., followed by addition of KOH, cooling and degassing of the emulsion.

#### Example 11

[0082] Broad spectrum sunscreen lotion containing 4% of a compound of Example 1.

wt%	Compound	INCI name	Supplier
<b>Part A</b>			
3.00	PARSOL MCX	Octyl methoxycinnamate	1
4.00	Product of Example 1		
3.00	PARSOL 500	4-Methylbenzylidene camphor	1
4.00	PARSOL 1789	4-tert. Butyl-4'-methoxy-dibenzoyl-methane	1
2.00	Glyceryl monostearate	Glyceryl stearate	
2.00	Cetyl alcohol extra	Cetyl alcohol	
2.00	GANEX V-220	PVP/Eicosene copolymer	10
4.00	CERAPHYL 375	Isostearyl neopentanoate	10
4.00	CERAPHYL 847	Octyldodecyl stearoyl stearate	10
2.00	AMPHISOL K	Potassium cetylphosphate	1
0.10	EDETA BD	Disodium EDTA	5
0.60	PHENONIP	Phenoxyethanol & Methyl-, Ethyl-, Propyl- & Butyl-paraben	6
<b>Part B</b>			
11.15	Deionized Water	Deionized Water	
50.00	CARBOPOL 934 1% solution	Carbomer	7
5.00	Propyleneglycol	1,2-Propanediol	
0.15	NIPAGIN M	Methylparaben	6
3.00	KOH (10%)	Potassium hydroxide	
q.s.	Perfume oil	Fragrance	

**[0083]** Part A was heated in a reactor to 85°C. When homogeneous Part B was added followed by addition of pre-heated KOH (75°C), cooling and degassing of the emulsion.

#### Example 12

**[0084]** Broad spectrum sunscreen cream containing 4% of the product described in Example 5. The cream has a low skin penetration quality:

wt%	Compound	INCI name	Supplier
Part A			
8.00	PARSOL SLX	Dimethico-diethylbenzalmalonate	1
4.00	Product of Ex. 5		
6.00	T-COTE 031	Titanium Dioxide & Dimethicone	3
10.00	ESTOL GTEH 3609	Trioctanoin	4
1.00	Cetyl Alcohol	Cetyl Alcohol	4
4.00	ESTOL GMM 3650	Glyceryl Myristate	
0.05	Butylated Hydroxytoluene	BHT	5
0.10	EDETA BD	Disodium EDTA	
0.60	PHENONIP	Phenoxyethanol & Methyl-, Ethyl-Propyl- & Butyl-paraben	6
2.00	AMPHISOL K	Potassium Cetyl Phosphate	1
Part B			
50.75	Deionized Water	Deionized Water	7
10.00	CARBOPOL 980 1 % sol'n	Carbomer 980	
3.00	Glycerin	Glycerin	
Part C			
0.50	KOH 10 % sol'n	Potassium Hydroxide	2
q.s	Perfume Oil	Fragrance	

[0085] Part A was heated to 85°C while stirring and then mixed for 30 sec. with a turbine at 8000 t/min. When homogeneous, Parts B and C were pre-heated to 75°C and added to Part A. The mixture was cooled to 40°C and Part D was added. The water loss was compensated and stirring was continued while cooling to ambient temperature. Then mixed 30 sec. with a turbine at 8000 t/min

#### Suppliers

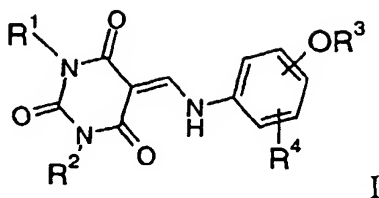
#### [0086]

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- 9) BERNEL Chemical Co. INC. Englewood NJ. USA

10) International Specialty Products ISP

## Claims

5 1. A compound of the general formula I



15 wherein

20  $R^1$  and  $R^2$  each independently are  $C_2$ - $C_{18}$  alkyl,  $C_5$ - $C_7$  cycloalkyl or  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen;

$R^3$  is hydrogen,  $C_1$ - $C_{18}$  alkyl or  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen, or is  $C_1$ - $C_{18}$  alkyl carbonyl,  $C_3$ - $C_{18}$  alkenyl,  $C_3$ - $C_{18}$  alkynyl or a group YS;

25  $R^4$  is hydrogen,  $C_1$ - $C_8$  alkyl or a group  $OR^3$ ;

Y is a linker group;

30 S is a silane-, an oligosiloxane- or a polysiloxane moiety.

2. A compound according to claim 1, wherein  $R^1$  and  $R^2$  are isopropyl.

3. A compound according to claim 1 or 2, wherein  $R^4$  is hydrogen or hydroxy.

35 4. A compound according to any one of claims 1 to 3, wherein  $OR^3$  is in the para position and  $R^3$  is 2-ethyl hexyl.

5. A compound according to any one of claims 1 to 3, wherein  $OR^3$  is in the para position and  $R^3$  is YS.

40 6. A compound according to claim 5, wherein Y is 3-propylene, 4-butylene, 2-propen-2-ylene, 2-propen-3-ylene or 3-buten-4-ylene.

7. A compound according to claim 5 or 6, wherein S is a group  $-SiR^5R^6R^7$ , wherein  $R^5$ ,  $R^6$  and  $R^7$  each independently are  $C_1$ - $C_6$  alkyl or phenyl.

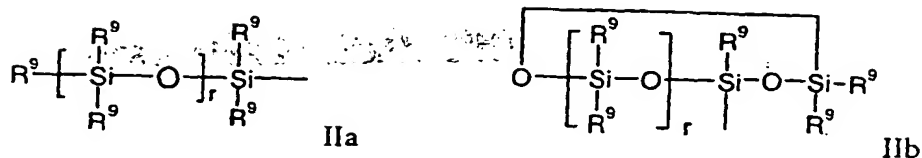
45 8. A compound according to claim 7, wherein S is trimethylsilane, triethylsilane, tripropylsilane, triisopropylsilane, dimethyl tert.butylsilane, dimethyl hexylsilane, triphenylsilane or dimethylphenylsilane.

9. A compound according to claim 5 or 6, wherein S is a group of the general formula  $-SiR^8_m(OSiR^{8'}_3)_n$  with  $m = 0, 1$  or  $2$ ;  $n = 1, 2$  or  $3$  and  $m+n = 3$ , wherein  $R^8$  and  $R^{8'}$  each independently are  $C_1$ - $C_6$  alkyl or phenyl.

50 10. A compound according to claim 9, wherein S is  $-SiCH_3(OSi(CH_3)_3)_2$ .

11. A compound according to claim 5 or 6, wherein S is a group of the general formula IIa or IIb

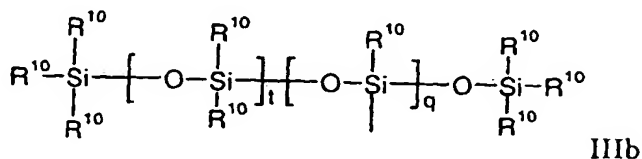
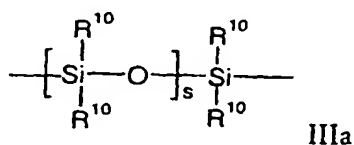
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wherein  $\text{R}^9$  is  $\text{C}_1\text{-C}_6$  alkyl or phenyl and  $r$  is integer from 1 to 9.

12. A compound according to claim 11, wherein  $\text{R}^9$  is methyl.

13. A compound according to claim 6, wherein S is a group of the general formulae IIIa or IIIb,



wherein

$\text{R}^{10}$  is  $\text{C}_1\text{-C}_6$  alkyl or phenyl;  
 $s$  is an integer from 4 to 250;  
 $t$  is an integer from 5 to 250;  
 $q$  is an integer from 1 to 30.

14. A compound according to claim 13, wherein

$\text{R}^{10}$  is  $\text{C}_1\text{-C}_4$  alkyl, preferably methyl;  
 $s$  is an integer from 4 to 150;  
 $t$  has a statistical mean value of about 60;  
 $q$  has a statistical mean value of about 4.

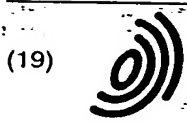
15. A light screening composition comprising a compound according to any one of claims 1 to 14.

16. A light screening composition according to claim 15, further comprising a conventional UV-A and/or UV-B screening agent.

17. The use of a compound according to any one of claims 1 to 16 as absorber for ultraviolet light.

18. The novel compounds, compositions and use substantially as described hereinbefore, especially with reference to the Examples.

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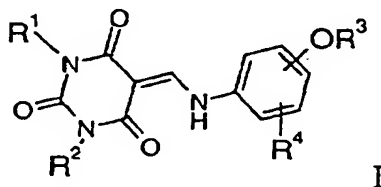
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(54) **Novel Pyrimidine-2,4,6-trione compounds**

(57) Compounds of the general formula I



wherein

- $R^1$  and  $R^2$  each independently are  $C_2$ - $C_{18}$  alkyl or  $C_5$ - $C_7$  cycloalkyl, or  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen;
- $R^3$  is hydrogen,  $C_1$ - $C_{18}$  alkyl or  $C_2$ - $C_{18}$  alkyl in which at least one methylene group is replaced by oxygen, or is  $C_1$ - $C_{18}$  alkyl carbonyl,  $C_3$ - $C_{18}$  alkenyl,  $C_3$ - $C_{18}$  alkynyl or a group YS;
- $R^4$  is hydrogen,  $C_1$ - $C_8$  alkyl or a group  $OR^3$ ;
- Y is a linker group;
- S is a silane-, an oligosiloxane- or a polysiloxane-moiety.

are effective in absorbing ultra violet radiation and can be used as sunscreen agents.

EP 1 081 140 A3



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Office

# EUROPEAN SEARCH REPORT

Application Number  
EP 00 11 8363

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	CHEMICAL ABSTRACTS, vol. 130, no. 24, 1999 Columbus, Ohio, US; abstract no. 325830s, page 653; XP002160718 * abstract * & JP 11 116556 A (CHEMIPRO KASEI) 27 April 1999 (1999-04-27) ---	1,15-18	C07D239/62 C07F7/10 C09K15/30 C07D239/60 C08G77/388
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A	US 5 561 179 A (V.BORZATTA) 1 October 1996 (1996-10-01) * column 19 - column 28; example 1 * ---	1,5, 15-18	TECHNICAL FIELDS SEARCHED (Int.Cl.7)
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P,X	CHEMICAL ABSTRACTS, vol. 131, no. 17, 1999 Columbus, Ohio, US; abstract no. 228730x, page 715; XP002160720 * abstract * & JP 11 263778 A (CHEMIPRO KASEI) 28 September 1999 (1999-09-28) -----	1,15-18	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 19 February 2001	Examiner Francois, J
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- &amp; : member of the same patent family, corresponding document</p>			

EPO FORM 1503 03 82 (P4/C01)



# ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
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